### **SECTION 1: Identification**

#### 1.1 GHS Product identifier

Product name Sulphuryl difluoride

#### 1.2 Other means of identification

Product number

Other names sulfuryl fluoride; Sulfur difluoride dioxide; Vikane

#### 1.3 Recommended use of the chemical and restrictions on use

**Identified uses**Industrial and scientific research uses.

Uses advised against no data available

### **SECTION 2: Hazard identification**

#### 2.1 Classification of the substance or mixture

Gases under pressure: Compressed gas Acute toxicity - Category 3, Inhalation

Specific target organ toxicity â€" repeated exposure, Category 2

Hazardous to the aquatic environment, short-term (Acute) - Category Acute 1

# 2.2 GHS label elements, including precautionary statements

#### Pictogram(s)









Signal word Danger

Hazard statement(s) H331 Toxic if inhaled

H373 May cause damage to organs through prolonged or repeated exposure

H400 Very toxic to aquatic life

Precautionary statement(s)

Prevention P261 Avoid breathing dust/fume/gas/mist/vapours/spray.

P271 Use only outdoors or in a well-ventilated area.
P260 Do not breathe dust/fume/gas/mist/vapours/spray.

P273 Avoid release to the environment.

**Response** P304+P340 IF INHALED: Remove person to fresh air and keep comfortable for breathing.

P316 Get emergency medical help immediately. P321 Specific treatment (see ... on this label). P319 Get medical help if you feel unwell.

P391 Collect spillage.

**Storage** P410+P403 Protect from sunlight. Store in a well-ventilated place.

P403+P233 Store in a well-ventilated place. Keep container tightly closed.

P405 Store locked up.

Disposal P501 Dispose of contents/container to an appropriate treatment and disposal facility in

accordance with applicable laws and regulations, and product characteristics at time of

disposal.

## 2.3 Other hazards which do not result in classification

no data available

# **SECTION 3: Composition/information on ingredients**

### 3.1 Substances

Chemical name	Common names and synonyms	CAS number	EC number	Concentration
Sulphuryl difluoride	Sulphuryl difluoride	2699-79-8	220-281-5	100%

### **SECTION 4: First-aid measures**

### 4.1 Description of necessary first-aid measures

## If inhaled

Fresh air, rest. Half-upright position. Refer for medical attention. See Notes.

#### Following skin contact

ON FROSTBITE: rinse with plenty of water, do NOT remove clothes.

#### Following eye contact

First rinse with plenty of water for several minutes (remove contact lenses if easily possible), then refer for medical attention.

#### **Following ingestion**

Rinse mouth with water. Do not induce vomiting. Never give anything by mouth to an unconscious person. Call a doctor or Poison Control Center immediately.

### 4.2 Most important symptoms/effects, acute and delayed

no data available

## 4.3 Indication of immediate medical attention and special treatment needed, if necessary

no data available

## **SECTION 5: Fire-fighting measures**

### 5.1 Suitable extinguishing media

Use dry chemical, carbon dioxide or alcohol-resistant foam.

## 5.2 Specific hazards arising from the chemical

Not combustible. Gives off irritating or toxic fumes (or gases) in a fire.

## 5.3 Special protective actions for fire-fighters

In case of fire in the surroundings, use appropriate extinguishing media. In case of fire: keep cylinder cool by spraying with water.

#### **SECTION 6: Accidental release measures**

### 6.1 Personal precautions, protective equipment and emergency procedures

NEVER direct water jet on liquid. Remove vapour with fine water spray. Personal protection: gas-tight chemical protection suit including self-contained breathing apparatus.

### **6.2** Environmental precautions

Prevent further spillage or leakage if it is safe to do so. Do not let the chemical enter drains. Discharge into the environment must be avoided.

### 6.3 Methods and materials for containment and cleaning up

Collect and arrange disposal. Keep the chemical in suitable and closed containers for disposal. Remove all sources of ignition. Use spark-proof tools and explosion-proof equipment. Adhered or collected material should be promptly disposed of, in accordance with appropriate laws and regulations.

## **SECTION 7: Handling and storage**

## 7.1 Precautions for safe handling

Handling in a well ventilated place. Wear suitable protective clothing. Avoid contact with skin and eyes. Avoid formation of dust and aerosols. Use non-sparking tools. Prevent fire caused by electrostatic discharge steam.

## 7.2 Conditions for safe storage, including any incompatibilities

Fireproof if in building. Keep in a well-ventilated room. Cool. Separated from food and feedstuffs.

# **SECTION 8: Exposure controls/personal protection**

## 8.1 Control parameters

#### Occupational Exposure limit values

TLV: 5 ppm as TWA; 10 ppm as STEL

#### **Biological limit values**

no data available

#### 8.2 Appropriate engineering controls

Ensure adequate ventilation. Handle in accordance with good industrial hygiene and safety practice. Set up emergency exits and the risk-elimination area.

#### 8.3 Individual protection measures, such as personal protective equipment (PPE)

# Eye/face protection

Wear face shield or eye protection in combination with breathing protection.

#### Skin protection

Cold-insulating gloves.

### Respiratory protection

Use ventilation, local exhaust or breathing protection.

#### Thermal hazards

# SECTION 9: Physical and chemical properties and safety characteristics

Physical state colourless odourless gas

Colourno data availableOdourno data availableMelting point/freezing point-121.4°CBoiling point or initial boiling point-55,2°C

and boiling range

Flammability no data available
Lower and upper explosion no data available

limit/flammability limit

Flash point no data available
Auto-ignition temperature no data available
Decomposition temperature no data available
pH no data available
Kinematic viscosity no data available

**Solubility** Solubility in water, ml/100ml: 4-5

Partition coefficient n-octanol/water no data available
Vapour pressure no data available
10700mmHg at 25°C

Density and/or relative density1.687g/cm3Relative vapour densityno data availableParticle characteristicsno data available

# **SECTION 10: Stability and reactivity**

### 10.1 Reactivity

no data available

### 10.2 Chemical stability

no data available

#### 10.3 Possibility of hazardous reactions

The gas is heavier than air and may accumulate in lowered spaces causing a deficiency of oxygen. Decomposes on heating. This produces toxic fumes including hydrogen fluoride and sulfur oxides.

#### 10.4 Conditions to avoid

no data available

## 10.5 Incompatible materials

no data available

### 10.6 Hazardous decomposition products

no data available

# **SECTION 11: Toxicological information**

#### Acute toxicity

- Oral: no data available
- Inhalation: no data available
- Dermal: no data available

### Skin corrosion/irritation

no data available

### Serious eye damage/irritation

no data available

# Respiratory or skin sensitization

no data available

### Germ cell mutagenicity

no data available

### Carcinogenicity

no data available

# Reproductive toxicity

#### STOT-single exposure

The substance is severely irritating to the respiratory tract. The substance may cause effects on the central nervous system. This may result in convulsions and respiratory failure. Inhalation of this gas may cause lung oedema. See Notes. Exposure could cause death. Rapid evaporation of the liquid may cause frostbite.

#### STOT-repeated exposure

no data available

#### Aspiration hazard

A harmful concentration of this gas in the air will be reached very quickly on loss of containment.

## **SECTION 12: Ecological information**

### 12.1 Toxicity

- Toxicity to fish: no data available
- Toxicity to daphnia and other aquatic invertebrates: no data available
- · Toxicity to algae: no data available
- Toxicity to microorganisms: no data available

## 12.2 Persistence and degradability

no data available

### 12.3 Bioaccumulative potential

no data available

### 12.4 Mobility in soil

no data available

#### 12.5 Other adverse effects

no data available

## **SECTION 13: Disposal considerations**

#### 13.1 Disposal methods

#### Product

The material can be disposed of by removal to a licensed chemical destruction plant or by controlled incineration with flue gas scrubbing. Do not contaminate water, foodstuffs, feed or seed by storage or disposal. Do not discharge to sewer systems.

## Contaminated packaging

Containers can be triply rinsed (or equivalent) and offered for recycling or reconditioning. Alternatively, the packaging can be punctured to make it unusable for other purposes and then be disposed of in a sanitary landfill. Controlled incineration with flue gas scrubbing is possible for combustible packaging materials.

# **SECTION 14: Transport information**

### 14.1 UN Number

ADR/RID: UN2191 (For reference only, please IMDG: UN2191 (For reference only, please IATA: UN2191 (For reference only, please check.)

# 14.2 UN Proper Shipping Name

ADR/RID: SULPHURYL FLUORIDE (For reference only, please check.)

IMDG: SULPHURYL FLUORIDE (For reference only, please check.)

IATA: SULPHURYL FLUORIDE (For reference only, please check.)

## 14.3 Transport hazard class(es)

ADR/RID: 2.3 (For reference only, please check.)

IMDG: 2.3 (For reference only, please check.)

IATA: 2.3 (For reference only, please check.)

## 14.4 Packing group, if applicable

ADR/RID: (For reference only, please check.) IMDG: (For reference only, please check.) IATA: (For reference only, please check.)

#### 14.5 Environmental hazards

ADR/RID: Yes IMDG: Yes IATA: Yes

## 14.6 Special precautions for user

no data available

## 14.7 Transport in bulk according to IMO instruments

# **SECTION 15: Regulatory information**

### 15.1 Safety, health and environmental regulations specific for the product in question

Chemical name	Common names and synonyms	CAS number	EC number
Sulphuryl difluoride	Sulphuryl difluoride	2699-79-8	220-281-5
European Inventory of Existing Commercial Chemical Substances (EINECS)			
EC Inventory			
United States Toxic Substances Control Act (TSCA) Inventory			
China Catalog of Hazardous chemicals 2015			
New Zealand Inventory of Chemicals (NZIoC)			
Philippines Inventory of Chemicals and Chemical Substances (PICCS)			
Vietnam National Chemical Inventory			
Chinese Chemical Inventory of Existing Chemical Substances (China IECSC)			
Korea Existing Chemicals List (KECL)			

### **SECTION 16: Other information**

#### Information on revision

**Creation Date** July 15, 2019 **Revision Date** July 15, 2019

#### Abbreviations and acronyms

- CAS: Chemical Abstracts Service
- ADR: European Agreement concerning the International Carriage of Dangerous Goods by Road
- RID: Regulation concerning the International Carriage of Dangerous Goods by Rail
- IMDG: International Maritime Dangerous Goods
- IATA: International Air Transportation Association
- TWA: Time Weighted Average
- STEL: Short term exposure limit
- LC50: Lethal Concentration 50%
- LD50: Lethal Dose 50%
- EC50: Effective Concentration 50%

#### References

- IPCS The International Chemical Safety Cards (ICSC), website: http://www.ilo.org/dyn/icsc/showcard.home
- HSDB Hazardous Substances Data Bank, website: https://toxnet.nlm.nih.gov/newtoxnet/hsdb.htm
- IARC International Agency for Research on Cancer, website: http://www.iarc.fr/
- eChemPortal The Global Portal to Information on Chemical Substances by OECD, website: http://www.echemportal.org/echemportal/index?pageID=0&request\_locale=en
- CAMEO Chemicals, website: http://cameochemicals.noaa.gov/search/simple
- ChemIDplus, website: http://chem.sis.nlm.nih.gov/chemidplus/chemidlite.jsp
- ERG Emergency Response Guidebook by U.S. Department of Transportation, website: http://www.phmsa.dot.gov/hazmat/library/erg
- Germany GESTIS-database on hazard substance, website: http://www.dguv.de/ifa/gestis/gestis-stoffdatenbank/index-2.jsp
- ECHA European Chemicals Agency, website: https://echa.europa.eu/

#### **Other Information**

Turn leaking cylinder with the leak up to prevent escape of gas in liquid state. The symptoms of lung oedema often do not become manifest until a few hours have passed and they are aggravated by physical effort. Rest and medical observation is therefore essential. High concentrations in the air cause a deficiency of oxygen with the risk of unconsciousness or death. Check oxygen content before entering area. There is no odour warning even when toxic concentrations are present.