SECTION 1: Identification

1.1 GHS Product identifier

Product name Arsenic acid

1.2 Other means of identification

Product number

Other names scorch;zotox;ARSENATE

1.3 Recommended use of the chemical and restrictions on use

Identified usesIntermediatesUses advised againstno data available

SECTION 2: Hazard identification

2.1 Classification of the substance or mixture

Acute toxicity - Category 3, Oral Acute toxicity - Category 3, Inhalation Carcinogenicity, Category 1A

Hazardous to the aquatic environment, long-term (Chronic) - Category Chronic 1

2.2 GHS label elements, including precautionary statements

Pictogram(s)







Signal word Danger

Hazard statement(s) H301 Toxic if swallowed

H331 Toxic if inhaled H350 May cause cancer H400 Very toxic to aquatic life

H410 Very toxic to aquatic life with long lasting effects

Precautionary statement(s)

Prevention P264 Wash ... thoroughly after handling.

P270 Do not eat, drink or smoke when using this product. P261 Avoid breathing dust/fume/gas/mist/vapours/spray. P271 Use only outdoors or in a well-ventilated area.

P203 Obtain, read and follow all safety instructions before use.

P280 Wear protective gloves/protective clothing/eye protection/face protection/hearing

protection/...

P273 Avoid release to the environment.

Response P301+P316 IF SWALLOWED: Get emergency medical help immediately.

P321 Specific treatment (see ... on this label).

P330 Rinse mouth.

P304+P340 IF INHALED: Remove person to fresh air and keep comfortable for breathing.

P316 Get emergency medical help immediately. P318 IF exposed or concerned, get medical advice.

P391 Collect spillage.

Storage P405 Store locked up.

P403+P233 Store in a well-ventilated place. Keep container tightly closed.

Disposal P501 Dispose of contents/container to an appropriate treatment and disposal facility in

accordance with applicable laws and regulations, and product characteristics at time of

disposal.

2.3 Other hazards which do not result in classification

no data available

SECTION 3: Composition/information on ingredients

3.1 Substances

Chemical name	Common names and synonyms	CAS number	EC number	Concentration
Arsenic acid	Arsenic acid	7778-39-4	231-901-9	100%

SECTION 4: First-aid measures

4.1 Description of necessary first-aid measures

If inhaled

Fresh air, rest. Refer for medical attention.

Following skin contact

Remove contaminated clothes. Rinse skin with plenty of water or shower. Refer for medical attention .

Following eye contact

First rinse with plenty of water for several minutes (remove contact lenses if easily possible), then refer for medical attention.

Following ingestion

Rinse mouth. Induce vomiting (ONLY IN CONSCIOUS PERSONS!). Rest. Refer for medical attention .

4.2 Most important symptoms/effects, acute and delayed

Excerpt from ERG Guide 154 [Substances - Toxic and/or Corrosive (Non-Combustible)]: TOXIC; inhalation, ingestion or skin contact with material may cause severe injury or death. Contact with molten substance may cause severe burns to skin and eyes. Avoid any skin contact. Effects of contact or inhalation may be delayed. Fire may produce irritating, corrosive and/or toxic gases. Runoff from fire control or dilution water may be corrosive and/or toxic and cause pollution. (ERG, 2016)

Ingestion cuases irritation of stomach, weakness, other gastrointestinal symptoms. Overdose can cause arsenic poisoning, but symptoms are delayed. (USCG, 1999)

4.3 Indication of immediate medical attention and special treatment needed, if necessary

Absorption, Distribution and Excretion

Systemic toxic effects have resulted from occupational accidents when arsenic acid ... splashed on workers, indicating that the skin is possible route of absorption for arsenic.

SECTION 5: Fire-fighting measures

5.1 Suitable extinguishing media

Personnel protection: ... Wear positive pressure self contained breathing apparatus when fighting fires involving arsenic acid. Arsenic acid solution; arsenic acid, solid

5.2 Specific hazards arising from the chemical

Excerpt from ERG Guide 154 [Substances - Toxic and/or Corrosive (Non-Combustible)]: Non-combustible, substance itself does not burn but may decompose upon heating to produce corrosive and/or toxic fumes. Some are oxidizers and may ignite combustibles (wood, paper, oil, clothing, etc.). Contact with metals may evolve flammable hydrogen gas. Containers may explode when heated. For electric vehicles or equipment, ERG Guide 147 (lithium ion batteries) or ERG Guide 138 (sodium batteries) should also be consulted. (ERG, 2016)

Excerpt from ERG Guide 154 [Substances - Toxic and/or Corrosive (Non-Combustible)]: Non-combustible, substance itself does not burn but may decompose upon heating to produce corrosive and/or toxic fumes. Some are oxidizers and may ignite combustibles (wood, paper, oil, clothing, etc.). Contact with metals may evolve flammable hydrogen gas. Containers may explode when heated. For electric vehicles or equipment, ERG Guide 147 (lithium ion batteries) or ERG Guide 138 (sodium batteries) should also be consulted. (ERG, 2016)

5.3 Special protective actions for fire-fighters

In case of fire in the surroundings, use appropriate extinguishing media.

SECTION 6: Accidental release measures

6.1 Personal precautions, protective equipment and emergency procedures

Evacuate danger area! Consult an expert! Personal protection: complete protective clothing including self-contained breathing apparatus. Do NOT let this chemical enter the environment. Collect leaking and spilled liquid in sealable plastic containers as far as possible. Absorb remaining liquid in sand or inert absorbent. Then store and dispose of according to local regulations.

6.2 Environmental precautions

Evacuate danger area! Consult an expert! Personal protection: complete protective clothing including self-contained breathing apparatus. Do NOT let this chemical enter the environment. Collect leaking and spilled liquid in sealable plastic containers as far as possible. Absorb remaining liquid in sand or inert absorbent. Then store and dispose of according to local regulations.

6.3 Methods and materials for containment and cleaning up

Prompt cleanup and removal are necessary. Control runoff and isolate discharged material for proper disposal.

SECTION 7: Handling and storage

7.1 Precautions for safe handling

Handling in a well ventilated place. Wear suitable protective clothing. Avoid contact with skin and eyes. Avoid formation of dust and aerosols. Use non-sparking tools. Prevent fire caused by electrostatic discharge steam.

7.2 Conditions for safe storage, including any incompatibilities

Store in an area without drain or sewer access. Separated from strong oxidants, strong bases, metals, strong reducing agents and food and feedstuffs. Do NOT store or transport in containers made from aluminium, copper, iron or zinc. Protect container against physical damage. Store in well ventilated area away from food or food products and combustible materials. Inorganic arsenic cmpd

8.1 Control parameters

Occupational Exposure limit values

TLV: (as As): 0.01 mg/m3, as TWA; A1 (confirmed human carcinogen); BEI issued.MAK: skin absorption (H); carcinogen category: 1; germ cell mutagen group: 3A

Biological limit values

no data available

8.2 Appropriate engineering controls

Ensure adequate ventilation. Handle in accordance with good industrial hygiene and safety practice. Set up emergency exits and the risk-elimination area.

8.3 Individual protection measures, such as personal protective equipment (PPE)

Eye/face protection

Wear face shield.

Skin protection

Protective gloves. Protective clothing.

Respiratory protection

Use closed system or ventilation.

Thermal hazards

no data available

SECTION 9: Physical and chemical properties and safety characteristics

Physical state Liquid

Colour WHITE TRANSLUCENT CRYSTALS /HEMIHYDRATE/

160 °C. Remarks:Loss of water.

Odourno data availableMelting point/freezing point $35 \, \hat{A}^{\circ} C$.

Boiling point or initial boiling point

and boiling range

no data available

Flammability no data available
Lower and upper explosion no data available

limit/flammability limit

Flash point no data available
Auto-ignition temperature no data available
Decomposition temperature no data available

pH WEAK ACIDIC PROPERTIES.

Kinematic viscosity no data available

Solubility 302 G/100 CC OF WATER @ 12.5 DEG C /HEMIHYDRATE/

Partition coefficient n-octanol/water no data available

Vapour pressure 55 mBar. Temperature:50 ŰC.;110 mBar. Temperature:65 ŰC.

Density and/or relative density2.5 g/cm³.Relative vapour densityno data availableParticle characteristicsno data available

SECTION 10: Stability and reactivity

10.1 Reactivity

5 mg/cu m (as AS); NIOSH considers arsenic (inorganic cmpd, as As) to be a potential occupational carcinogen. Arsenic (inorganic cmpd, as As)

Decomposes on heating. This produces toxic and corrosive fumes. The substance is a strong oxidant. It reacts with combustible and reducing materials. The substance is a medium strong acid. Attacks metals. This produces toxic and flammable arsine (see ICSC 0222).

10.2 Chemical stability

The pH of aqueous solutions appears to be a major factor in the relative stability. ... Pentavalent inorganic arsenic ... is relatively stable at neutral or alkaline pH but undergoes reduction with decreasing pH. Pentavalent inorganic arsenic

10.3 Possibility of hazardous reactions

ARSENIC ACID SOLUTION may evolve very toxic gaseous arsine (AsH3) from contact with active metals, such as zinc and aluminum [Inorganic Chemicals Handbook, I, p. 175]. When heated to decomposition, it produces toxic fumes of metallic arsenic [Sax, 9th ed., p. 271]. The solution is weakly acidic and a weak oxidizing agent. Reacts with alkalis to generate some heat and precipitate arsenates.

10.4 Conditions to avoid

no data available

10.5 Incompatible materials

Arsenic acid is a strong oxidant and will oxidize I(-1) ion to I(-3).

10.6 Hazardous decomposition products

When heated to decomposition it emits toxic fumes of arsenic.

SECTION 11: Toxicological information

Acute toxicity

- Oral: LD50 Rat oral 48 mg/kg
- Inhalation: no data available
- Dermal: no data available

Skin corrosion/irritation

no data available

Serious eye damage/irritation

no data available

Respiratory or skin sensitization

no data available

Germ cell mutagenicity

no data available

Carcinogenicity

Classification of carcinogenicity: 1) evidence in humans: sufficient; 2) evidence in animals: limited. Overall summary evaluation of carcinogenic risk to humans is Group 1: Carcinogenic to humans. NOTE: This evaluation applies to the group of chemicals as a whole and not necessarily to all individual chemicals within the group. Arsenic and arsenic compounds

Reproductive toxicity

no data available

STOT-single exposure

The substance is corrosive to the eyes, skin and respiratory tract. The substance may cause effects on the blood, cardiovascular system, gastrointestinal tract, liver and nervous system. The effects may be delayed. See Notes.

STOT-repeated exposure

The substance may have effects on the peripheral nervous system, skin and cardiovascular system. This may result in polyneuropathy and skin lesions. This substance is carcinogenic to humans.

Aspiration hazard

A harmful contamination of the air can be reached very quickly on evaporation of this substance at 20°C on spraying.

SECTION 12: Ecological information

12.1 Toxicity

- Toxicity to fish: LC50 Oncorhynchus mykiss (previous name: Salmo gairdneri) ca. 67.5 mg/L 96 h.
- Toxicity to daphnia and other aquatic invertebrates: LOEC Strongylocentrotus purpuratus 11 Âμg/L 48 h.
- $\bullet~$ Toxicity to algae: EC50 Ankistrodesmus falcatus 0.256 mg/L 14 d.
- Toxicity to microorganisms: EC10 Pseudomonas putida 18 mg arsenate/L 16 h.

12.2 Persistence and degradability

no data available

12.3 Bioaccumulative potential

no data available

12.4 Mobility in soil

no data available

12.5 Other adverse effects

no data available

SECTION 13: Disposal considerations

13.1 Disposal methods

Product

The material can be disposed of by removal to a licensed chemical destruction plant or by controlled incineration with flue gas scrubbing. Do not contaminate water, foodstuffs, feed or seed by storage or disposal. Do not discharge to sewer systems.

Contaminated packaging

Containers can be triply rinsed (or equivalent) and offered for recycling or reconditioning. Alternatively, the packaging can be punctured to make it unusable for other purposes and then be disposed of in a sanitary landfill. Controlled incineration with flue gas scrubbing is possible for combustible packaging materials.

SECTION 14: Transport information

14.1 UN Number

ADR/RID: UN1553 (For reference only, please IMDG: UN1553 (For reference only, please IATA: UN1553 (For reference only, please check.) check.)

14.2 UN Proper Shipping Name

ADR/RID: ARSENIC ACID, LIQUID (For reference only, please check.)

IMDG: ARSENIC ACID, LIQUID (For reference only, please check.)

IATA: ARSENIC ACID, LIQUID (For reference only, please check.)

14.3 Transport hazard class(es)

ADR/RID: 6.1 (For reference only, please check.)

IMDG: 6.1 (For reference only, please check.)

IATA: 6.1 (For reference only, please check.)

14.4 Packing group, if applicable

ADR/RID: I (For reference only, please check.) IMDG: I (For reference only, please check.) IATA: I (For reference only, please check.)

14.5 Environmental hazards

ADR/RID: Yes IMDG: Yes IATA: Yes

14.6 Special precautions for user

no data available

14.7 Transport in bulk according to IMO instruments

no data available

SECTION 15: Regulatory information

15.1 Safety, health and environmental regulations specific for the product in question

Chemical name	Common names and synonyms	CAS number	EC number
Arsenic acid	Arsenic acid	7778-39-4	231-901-9
European Inventory of Existing Commercial Chemical Substances (EINECS)			
EC Inventory			
United States Toxic Substances Control Act (TSCA) Inventory			
China Catalog of Hazardous chemicals 2015			
New Zealand Inventory of Chemicals (NZIoC)			
Philippines Inventory of Chemicals and Chemical Substances (PICCS)			
Vietnam National Chemical Inventory			
Chinese Chemical Inventory of Existing Chemical Substances (China IECSC)			
Korea Existing Chemicals List (KECL)			

SECTION 16: Other information

Information on revision

Creation DateJuly 15, 2019Revision DateJuly 15, 2019

Abbreviations and acronyms

- CAS: Chemical Abstracts Service
- ADR: European Agreement concerning the International Carriage of Dangerous Goods by Road
- RID: Regulation concerning the International Carriage of Dangerous Goods by Rail
- IMDG: International Maritime Dangerous Goods
- IATA: International Air Transportation Association
- TWA: Time Weighted Average
- STEL: Short term exposure limit
- LC50: Lethal Concentration 50%
- LD50: Lethal Dose 50%
- EC50: Effective Concentration 50%

References

- IPCS The International Chemical Safety Cards (ICSC), website: http://www.ilo.org/dyn/icsc/showcard.home
- HSDB Hazardous Substances Data Bank, website: https://toxnet.nlm.nih.gov/newtoxnet/hsdb.htm
- IARC International Agency for Research on Cancer, website: http://www.iarc.fr/
- eChemPortal The Global Portal to Information on Chemical Substances by OECD, website: http://www.echemportal.org/echemportal/index?pageID=0&request locale=en
- CAMEO Chemicals, website: http://cameochemicals.noaa.gov/search/simple
- ChemIDplus, website: http://chem.sis.nlm.nih.gov/chemidplus/chemidlite.jsp
- ERG Emergency Response Guidebook by U.S. Department of Transportation, website: http://www.phmsa.dot.gov/hazmat/library/erg
- Germany GESTIS-database on hazard substance, website: http://www.dguv.de/ifa/gestis/gestis-stoffdatenbank/index-2.jsp
- ECHA European Chemicals Agency, website: https://echa.europa.eu/

Other Information

Depending on the degree of exposure, periodic medical examination is suggested. The symptoms of acute poisoning do not become manifest until hours. Do NOT take working clothes home. Another CAS number used is 1327-52-2. Arsenic acid hemihydrate has CAS number 7774-41-6. Arsenic acid (hemihydrate) is a colourless to white hygroscopic solid. It decomposes at $160 \text{\AA}^{\circ}\text{C}$ (loss of water); melting point $35 \text{\AA}^{\circ}\text{C}$; relative density 2-2.5; very soluble in water. UN number 1554.