# **SECTION 1: Identification**

#### 1.1 GHS Product identifier

Product name Cobalt sulphate

#### 1.2 Other means of identification

Product number

Other names COBATOUSSULPHATE; cobalt(ii) sulphate; cobalt(2+) sulfate

#### 1.3 Recommended use of the chemical and restrictions on use

Identified uses Agricultural chemicals (non-pesticidal), Corrosion inhibitors and anti-scaling

agents, Intermediates, Plating agents and surface treating agents, Processing aids, not otherwise

listed, Processing aids, specific to petroleum production

Uses advised against no data available

### **SECTION 2: Hazard identification**

#### 2.1 Classification of the substance or mixture

Acute toxicity - Category 4, Oral Skin sensitization, Category 1 Respiratory sensitization, Category 1 Germ cell mutagenicity, Category 2

Hazardous to the aquatic environment, short-term (Acute) - Category Acute 1 Hazardous to the aquatic environment, long-term (Chronic) - Category Chronic 1

Carcinogenicity, Category 1B Reproductive toxicity, Category 1B

# 2.2 GHS label elements, including precautionary statements

### Pictogram(s)







Signal word Danger

Hazard statement(s) H302 Harmful if swallowed

H317 May cause an allergic skin reaction

H334 May cause allergy or asthma symptoms or breathing difficulties if inhaled

H341 Suspected of causing genetic defects

H410 Very toxic to aquatic life with long lasting effects

Precautionary statement(s)

**Prevention** P264 Wash ... thoroughly after handling.

P270 Do not eat, drink or smoke when using this product. P261 Avoid breathing dust/fume/gas/mist/vapours/spray.

P272 Contaminated work clothing should not be allowed out of the workplace. P280 Wear protective gloves/protective clothing/eye protection/face protection/hearing

protection/...

P284 [In case of inadequate ventilation] wear respiratory protection. P203 Obtain, read and follow all safety instructions before use.

P273 Avoid release to the environment.

**Response** P301+P317 IF SWALLOWED: Get medical help.

P330 Rinse mouth.

P302+P352 IF ON SKIN: Wash with plenty of water/... P333+P317 If skin irritation or rash occurs: Get medical help.

P321 Specific treatment (see ... on this label).

 $P362 + P364 \ Take \ off \ contaminated \ clothing \ and \ wash \ it \ before \ reuse.$ 

P304+P340 IF INHALED: Remove person to fresh air and keep comfortable for breathing. P342+P316 If experiencing respiratory symptoms: Get emergency medical help immediately.

P318 IF exposed or concerned, get medical advice.

P391 Collect spillage.

Storage P405 Store locked up.

**Disposal** P501 Dispose of contents/container to an appropriate treatment and disposal facility in

accordance with applicable laws and regulations, and product characteristics at time of

disposal.

### 2.3 Other hazards which do not result in classification

no data available

# **SECTION 3: Composition/information on ingredients**

#### 3.1 Substances

Chemical name	Common names and synonyms	CAS number	EC number	Concentration
Cobalt sulphate	Cobalt sulphate	10124-43-3	233-334-2	100%

### **SECTION 4: First-aid measures**

# 4.1 Description of necessary first-aid measures

#### If inhaled

Fresh air, rest. Refer for medical attention.

#### Following skin contact

Remove contaminated clothes. Rinse and then wash skin with water and soap. Seek medical attention if you feel unwell.

#### Following eye contact

First rinse with plenty of water for several minutes (remove contact lenses if easily possible), then refer for medical attention.

### **Following ingestion**

Rinse mouth. Give one or two glasses of water to drink. Refer for medical attention.

### 4.2 Most important symptoms/effects, acute and delayed

Inhalation causes shortness of breath and coughing; permanent disability may occur. Ingestion causes pain and vomiting. Contact with eyes or skin causes irritation. (USCG, 1999)

### 4.3 Indication of immediate medical attention and special treatment needed, if necessary

#### Absorption, Distribution and Excretion

Sheep given a single dose of hydrated cobalt sulfate were killed. ... The livers contained 400-1200 ppm of cobalt. In cattle dying after a massive overdose of hydrated cobalt sulfate livers contained 5-300 ppm, kidneys 30-200 ppm. Hydrated cobalt sulfate

# **SECTION 5: Fire-fighting measures**

### 5.1 Suitable extinguishing media

In case of fire in the surroundings, use appropriate extinguishing media.

# 5.2 Specific hazards arising from the chemical

Special Hazards of Combustion Products: Toxic cobalt oxide fumes may form in fire. (USCG, 1999)

# 5.3 Special protective actions for fire-fighters

In case of fire in the surroundings, use appropriate extinguishing media.

### **SECTION 6: Accidental release measures**

### 6.1 Personal precautions, protective equipment and emergency procedures

Personal protection: particulate filter respirator adapted to the airborne concentration of the substance. Do NOT let this chemical enter the environment. Sweep spilled substance into covered containers. If appropriate, moisten first to prevent dusting. Carefully collect remainder. Then store and dispose of according to local regulations.

### 6.2 Environmental precautions

Personal protection: particulate filter respirator adapted to the airborne concentration of the substance. Do NOT let this chemical enter the environment. Sweep spilled substance into covered containers. If appropriate, moisten first to prevent dusting. Carefully collect remainder. Then store and dispose of according to local regulations.

## 6.3 Methods and materials for containment and cleaning up

Collect and arrange disposal. Keep the chemical in suitable and closed containers for disposal. Remove all sources of ignition. Use spark-proof tools and explosion-proof equipment. Adhered or collected material should be promptly disposed of, in accordance with appropriate laws and regulations.

# **SECTION 7: Handling and storage**

## 7.1 Precautions for safe handling

Handling in a well ventilated place. Wear suitable protective clothing. Avoid contact with skin and eyes. Avoid formation of dust and aerosols. Use non-sparking tools. Prevent fire caused by electrostatic discharge steam.

# 7.2 Conditions for safe storage, including any incompatibilities

Separated from strong oxidants. Store in an area without drain or sewer access. Provision to contain effluent from fire extinguishing. Separated from strong oxidants.

# **SECTION 8: Exposure controls/personal protection**

# 8.1 Control parameters

### Occupational Exposure limit values

TLV: (as Co): 0.02 mg/m3, as TWA; A3 (confirmed animal carcinogen with unknown relevance to humans); BEI issued.MAK: (as Co, inhalable fraction): skin absorption (H); sensitization of respiratory tract and skin (SAH); carcinogen category: 2; germ cell mutagen group: 3A

#### **Biological limit values**

no data available

# 8.2 Appropriate engineering controls

Ensure adequate ventilation. Handle in accordance with good industrial hygiene and safety practice. Set up emergency exits and the risk-elimination area

# 8.3 Individual protection measures, such as personal protective equipment (PPE)

#### Eye/face protection

Wear safety goggles or eye protection in combination with breathing protection if powder.

#### Skin protection

Protective gloves. Protective clothing.

#### Respiratory protection

Use local exhaust or breathing protection.

#### Thermal hazards

no data available

# SECTION 9: Physical and chemical properties and safety characteristics

Physical state Solid. Crystalline.

Colour Rose.

Odourno data availableMelting point/freezing point $> 700 \text{ Å}^{\circ}\text{C}$ .

**Boiling point or initial boiling point** 

and boiling range

330°C at 760mmHg

Flammability Not combustible. Gives off irritating or toxic fumes (or gases) in a fire.

Lower and upper explosion no data available

limit/flammability limit

Flash point no data available
Auto-ignition temperature no data available
Decomposition temperature 735°C
pH no data available

Kinematic viscosity no data available

Solubility 38.3 G/100 ML WATER @ 25 DEG C

Partition coefficient n-octanol/water no data available

**Vapour pressure** 3.35E-05mmHg at 25°C **Density and/or relative density** 3.71. Temperature:25 °C.

Relative vapour density no data available

Particle characteristics no data available

# **SECTION 10: Stability and reactivity**

#### 10.1 Reactivity

Decomposes at  $735 \text{Å}^{\circ}\text{C}$ . This produces toxic fumes of sulfur oxides. The dust reacts with strong oxidants. This generates fire and explosion hazard.

# 10.2 Chemical stability

Stable to 708 deg C.

# 10.3 Possibility of hazardous reactions

Acidic salts, such as COBALT SULFATE, are generally soluble in water. The resulting solutions contain moderate concentrations of hydrogen ions and have pH's of less than 7.0. They react as acids to neutralize bases. These neutralizations generate heat, but less or far less than is generated by neutralization of inorganic acids, inorganic oxoacids, and carboxylic acid. They usually do not react as either oxidizing agents or reducing agents but such behavior is not impossible. Many of these compounds catalyze organic reactions.

#### 10.4 Conditions to avoid

no data available

### 10.5 Incompatible materials

Reacts as a dust with strong oxidants causing fire and explosion hazard.

# 10.6 Hazardous decomposition products

The substance decomposes on heating to 735C, producing toxic fumes of sulfur oxides

# **SECTION 11: Toxicological information**

#### Acute toxicity

- Oral: LD50 rat (male/female) 768 mg/kg bw. Remarks: This is the LD50 for the cobalt compound tested.
- · Inhalation: no data available
- Dermal: LD50 rat (male/female) > 2 000 mg/kg bw.

#### Skin corrosion/irritation

no data available

### Serious eye damage/irritation

no data available

#### Respiratory or skin sensitization

no data available

#### Germ cell mutagenicity

no data available

#### Carcinogenicity

Cobaltous Sulfate: reasonably anticipated to be a human carcinogen.

#### Reproductive toxicity

no data available

### STOT-single exposure

The is severely irritating to the respiratory tract. The substance is irritating to the eyes.

#### STOT-repeated exposure

Repeated or prolonged contact may cause skin sensitization. Repeated or prolonged inhalation may cause asthma. Ingestion may cause effects on the heart, bone marrow and thyroid. This substance is possibly carcinogenic to humans. Animal tests show that this substance possibly causes toxic effects upon human reproduction.

#### Aspiration hazard

A harmful concentration of airborne particles can be reached quickly when dispersed, especially if powdered.

# **SECTION 12: Ecological information**

### 12.1 Toxicity

- Toxicity to fish: LC50 Pimephales promelas 54.1 mg/L 96 h.
- Toxicity to daphnia and other aquatic invertebrates: NOEC Chironomus tentans 72.3 mg/L 96 h.
- Toxicity to algae: NOEC Dunaliella tertiolecta 4 671.8 Âμg/L 96 h.
- Toxicity to microorganisms: EC10 activated sludge 3.73 mg/L 30 min.

# 12.2 Persistence and degradability

no data available

# 12.3 Bioaccumulative potential

Food Chain Concn Potential: Bioconcentration of 200-1000 fold only under constant exposure. Not significant in spill conditions. Heptahydrate

#### 12.4 Mobility in soil

no data available

### 12.5 Other adverse effects

no data available

# **SECTION 13: Disposal considerations**

# 13.1 Disposal methods

#### **Product**

The material can be disposed of by removal to a licensed chemical destruction plant or by controlled incineration with flue gas scrubbing. Do not contaminate water, foodstuffs, feed or seed by storage or disposal. Do not discharge to sewer systems.

### Contaminated packaging

Containers can be triply rinsed (or equivalent) and offered for recycling or reconditioning. Alternatively, the packaging can be punctured to make it unusable for other purposes and then be disposed of in a sanitary landfill. Controlled incineration with flue gas scrubbing is possible for combustible packaging materials.

# **SECTION 14: Transport information**

### 14.1 UN Number

ADR/RID: UN3077 (For reference only, please IMDG: UN3077 (For reference only, please IATA: UN3077 (For reference only, please check.)

# 14.2 UN Proper Shipping Name

ADR/RID: ENVIRONMENTALLY IMDG: ENVIRONMENTALLY IATA: ENVIRONMENTALLY HAZARDOUS SUBSTANCE, SOLID, N.O.S. (For reference only, please check.)

IATA: ENVIRONMENTALLY HAZARDOUS SUBSTANCE, SOLID, HAZARDOUS SUBSTANCE, SOLID, N.O.S. (For reference only, please check.)

### 14.3 Transport hazard class(es)

ADR/RID: 9 (For reference only, please check.)

IMDG: 9 (For reference only, please check.)

IATA: 9 (For reference only, please check.)

# 14.4 Packing group, if applicable

ADR/RID: III (For reference only, please check.)

IMDG: III (For reference only, please check.)

IATA: III (For reference only, please check.)

### 14.5 Environmental hazards

ADR/RID: Yes IMDG: Yes IATA: Yes

# 14.6 Special precautions for user

no data available

### 14.7 Transport in bulk according to IMO instruments

no data available

# **SECTION 15: Regulatory information**

# 15.1 Safety, health and environmental regulations specific for the product in question

Chemical name	Common names and synonyms	CAS number	EC number
Cobalt sulphate	Cobalt sulphate	10124-43-3	233-334-2
<b>European Inventory of Exist</b>	ing Commercial Chemical Substances (EINECS)		Listed.
EC Inventory			Listed.
United States Toxic Substances Control Act (TSCA) Inventory			Listed.
China Catalog of Hazardous chemicals 2015			Listed.
New Zealand Inventory of Cl	nemicals (NZIoC)		Listed.
Philippines Inventory of Che	micals and Chemical Substances (PICCS)		Listed.
Vietnam National Chemical	Inventory		Listed.
Chinese Chemical Inventory	of Existing Chemical Substances (China IECSC)		Listed.
Korea Existing Chemicals Li	st (KECL)		Listed.

# **SECTION 16: Other information**

# Information on revision

Creation DateJuly 15, 2019Revision DateJuly 15, 2019

### Abbreviations and acronyms

- CAS: Chemical Abstracts Service
- ADR: European Agreement concerning the International Carriage of Dangerous Goods by Road
- RID: Regulation concerning the International Carriage of Dangerous Goods by Rail
- IMDG: International Maritime Dangerous Goods
- IATA: International Air Transportation Association
- TWA: Time Weighted Average
- STEL: Short term exposure limit
- LC50: Lethal Concentration 50%
- LD50: Lethal Dose 50%
- EC50: Effective Concentration 50%

#### References

- IPCS The International Chemical Safety Cards (ICSC), website: http://www.ilo.org/dyn/icsc/showcard.home
- HSDB Hazardous Substances Data Bank, website: https://toxnet.nlm.nih.gov/newtoxnet/hsdb.htm

- IARC International Agency for Research on Cancer, website: http://www.iarc.fr/
- eChemPortal The Global Portal to Information on Chemical Substances by OECD, website: http://www.echemportal.org/echemportal/index?pageID=0&request\_locale=en
- CAMEO Chemicals, website: http://cameochemicals.noaa.gov/search/simple
- $\bullet \quad ChemIDplus, \ website: \ http://chem.sis.nlm.nih.gov/chemidplus/chemidlite.jsp$
- ERG Emergency Response Guidebook by U.S. Department of Transportation, website: http://www.phmsa.dot.gov/hazmat/library/erg
- Germany GESTIS-database on hazard substance, website: http://www.dguv.de/ifa/gestis/gestis-stoffdatenbank/index-2.jsp
- ECHA European Chemicals Agency, website: https://echa.europa.eu/

### Other Information

The symptoms of asthma often do not become manifest until a few hours have passed and they are aggravated by physical effort. Rest and medical observation are therefore essential. Anyone who has shown symptoms of asthma due to this substance should avoid all further contact. Depending on the degree of exposure, periodic medical examination is suggested. Do NOT take working clothes home.