## **SECTION 1: Identification**

#### 1.1 GHS Product identifier

**Product name** Diacetyl peroxide

#### 1.2 Other means of identification

Product number

Other names Peroxide, diacetyl; Acetyl peroxide; Peroyl A

#### 1.3 Recommended use of the chemical and restrictions on use

**Identified uses**Industrial and scientific research uses.

Uses advised against no data available

## **SECTION 2: Hazard identification**

#### 2.1 Classification of the substance or mixture

no data available

# 2.2 GHS label elements, including precautionary statements

Pictogram(s)no data availableSignal wordno data availableHazard statement(s)no data available

**Precautionary statement(s)** 

Preventionno data availableResponseno data availableStorageno data availableDisposalno data available

#### 2.3 Other hazards which do not result in classification

no data available

# **SECTION 3: Composition/information on ingredients**

#### 3.1 Substances

Chemical name	Common names and synonyms	CAS number	EC number	Concentration
Diacetyl peroxide	Diacetyl peroxide	110-22-5	203-748-8	100%

## **SECTION 4: First-aid measures**

## 4.1 Description of necessary first-aid measures

#### If inhaled

Move the victim into fresh air. If breathing is difficult, give oxygen. If not breathing, give artificial respiration and consult a doctor immediately. Do not use mouth to mouth resuscitation if the victim ingested or inhaled the chemical.

#### Following skin contact

Take off contaminated clothing immediately. Wash off with soap and plenty of water. Consult a doctor.

#### Following eve contact

Rinse with pure water for at least 15 minutes. Consult a doctor.

#### **Following ingestion**

Rinse mouth with water. Do not induce vomiting. Never give anything by mouth to an unconscious person. Call a doctor or Poison Control Center immediately.

# 4.2 Most important symptoms/effects, acute and delayed

Excerpt from ERG Guide 148 [Organic Peroxides (Heat and Contamination Sensitive / Temperature Controlled)]: Fire may produce irritating, corrosive and/or toxic gases. Ingestion or contact (skin, eyes) with substance may cause severe injury or burns. Runoff from fire control or dilution water may cause pollution. (ERG, 2016)

## 4.3 Indication of immediate medical attention and special treatment needed, if necessary

Basic treatment: Establish a patent airway. Suction if necessary. Watch for signs of respiratory insufficiency and assist ventilations if necessary. Administer oxygen by nonrebreather mask at 10 to 15 L/min. Monitor for pulmonary edema and treat if necessary. Monitor for shock and treat if necessary. Anticipate seizures and treat if necessary. For eye contamination, flush eyes immediately with water. Irrigate each eye continuously with normal saline during transport. Do not use emetics. For ingestion, rinse mouth and administer 5 ml/kg up to 200 ml of water for dilution if the patient can swallow, has a strong gag reflex, and does not drool .. Do not attempt to neutralize because of exothermic reaction. Cover skin burns with dry, sterile dressings after decontamination. Organic peroxides

## 5.1 Suitable extinguishing media

Fight...with water from explosion-resistant location. in advanced or massive fires...area should be evacuated. if fire occurs in vicinity of this material water should be used to keep containers cool. clean-up & salvage operations should not be attempted until all...cooled completely. 25% soln

#### 5.2 Specific hazards arising from the chemical

Excerpt from ERG Guide 148 [Organic Peroxides (Heat and Contamination Sensitive / Temperature Controlled)]: May explode from heat, contamination or loss of temperature control. These materials are particularly sensitive to temperature rises. Above a given "Control Temperature" they decompose violently and catch fire. May ignite combustibles (wood, paper, oil, clothing, etc.). May ignite spontaneously if exposed to air. May be ignited by heat, sparks or flames. May burn rapidly with flare-burning effect. Containers may explode when heated. Runoff may create fire or explosion hazard. (ERG, 2016)

## 5.3 Special protective actions for fire-fighters

Wear self-contained breathing apparatus for firefighting if necessary.

#### **SECTION 6: Accidental release measures**

#### 6.1 Personal precautions, protective equipment and emergency procedures

Avoid dust formation. Avoid breathing mist, gas or vapours. Avoid contacting with skin and eye. Use personal protective equipment. Wear chemical impermeable gloves. Ensure adequate ventilation. Remove all sources of ignition. Evacuate personnel to safe areas. Keep people away from and upwind of spill/leak.

## **6.2** Environmental precautions

Prevent further spillage or leakage if it is safe to do so. Do not let the chemical enter drains. Discharge into the environment must be avoided

# 6.3 Methods and materials for containment and cleaning up

In event of spillage...spilled material should be absorbed...with noncombustible absorbent, such as vermiculite. sweep up & place in plastic container for immediate disposal. do not use spark-generating metals or cellulosic materials (paper, wood, etc) for sweeping up or handling spilled material. 25% soln

# **SECTION 7: Handling and storage**

## 7.1 Precautions for safe handling

Handling in a well ventilated place. Wear suitable protective clothing. Avoid contact with skin and eyes. Avoid formation of dust and aerosols. Use non-sparking tools. Prevent fire caused by electrostatic discharge steam.

# 7.2 Conditions for safe storage, including any incompatibilities

Protect against physical damage. store only in detached, isolated noncombustible building erected for this purpose & used exclusively for this material. do not store solid or paste peroxides in same building... no electrical installation, open flames or other sources of ignition permitted in storage building. 25% soln

# **SECTION 8: Exposure controls/personal protection**

## 8.1 Control parameters

#### Occupational Exposure limit values

no data available

#### **Biological limit values**

no data available

# 8.2 Appropriate engineering controls

Ensure adequate ventilation. Handle in accordance with good industrial hygiene and safety practice. Set up emergency exits and the risk-elimination area.

# 8.3 Individual protection measures, such as personal protective equipment (PPE)

### Eye/face protection

Wear tightly fitting safety goggles with side-shields conforming to EN 166(EU) or NIOSH (US).

## Skin protection

Wear fire/flame resistant and impervious clothing. Handle with gloves. Gloves must be inspected prior to use. Wash and dry hands. The selected protective gloves have to satisfy the specifications of EU Directive 89/686/EEC and the standard EN 374 derived from it.

#### Respiratory protection

If the exposure limits are exceeded, irritation or other symptoms are experienced, use a full-face respirator.

## Thermal hazards

no data available

Physical state The pure compound consists of colorless crystals melting at 30ŰC. Soluble in alcohol and

ether. Often transported as a 25% solution in dimethyl phthalate for use as an initiator and

catalyst.

ColourColorless crystalsOdourStrong, pungent odor

Melting point/freezing point 30ŰC

Boiling point or initial boiling point

and boiling range

121.4°C at 760mmHg

Flammability no data available
Lower and upper explosion no data available

limit/flammability limit

Flash point
32.2°C
Auto-ignition temperature
no data available
Decomposition temperature
no data available
pH
no data available
Kinematic viscosity
no data available

**Solubility** Slightly soluble in cold water

**Partition coefficient n-octanol/water** no data available **Vapour pressure** 14.6mmHg at 25°C

Density and/or relative density1.163g/cm3Relative vapour densityno data availableParticle characteristicsno data available

# **SECTION 10: Stability and reactivity**

### 10.1 Reactivity

No rapid reaction with air. No rapid reaction with water.

## 10.2 Chemical stability

Highly unstable org peroxides

## 10.3 Possibility of hazardous reactions

DANGEROUS, BY SPONTANEOUS CHEMICAL REACTION. A POWERFUL OXIDIZING AGENT; CAN CAUSE IGNITION OF ORGANIC MATERIALS ON CONTACT. Pure DIACETYL PEROXIDE, a solid, presents a severe explosion hazard. Unpredictably shock sensitive. Five grains detonated violently while being removed from an ice chest [Chem. Eng. News 26:3197(1948)]. May explode violently in contact with ether or any volatile solvent. A 5-gram portion in ether detonated while being carried [Chem. Eng. News 27:175(1949)]. The 25% solution in dimethyl phthalate is less dangerous, but still a strong oxidizing agent that presents a moderate fire risk [Hawley].

## 10.4 Conditions to avoid

no data available

#### 10.5 Incompatible materials

Highly dangerous...it will react with water or steam to produce heat; can react vigorously with reducing materials; can emit toxic fumes on contact with acid or acid fumes.

### 10.6 Hazardous decomposition products

no data available

# **SECTION 11: Toxicological information**

# Acute toxicity

- Oral: no data available
- Inhalation: no data available
- Dermal: no data available

#### Skin corrosion/irritation

no data available

#### Serious eye damage/irritation

no data available

#### Respiratory or skin sensitization

no data available

#### Germ cell mutagenicity

no data available

#### Carcinogenicity

no data available

#### Reproductive toxicity

no data available

#### STOT-single exposure

no data available

#### STOT-repeated exposure

no data available

#### **Aspiration hazard**

no data available

# **SECTION 12: Ecological information**

#### 12.1 Toxicity

- · Toxicity to fish: no data available
- · Toxicity to daphnia and other aquatic invertebrates: no data available
- Toxicity to algae: no data available
- Toxicity to microorganisms: no data available

# 12.2 Persistence and degradability

no data available

#### 12.3 Bioaccumulative potential

no data available

## 12.4 Mobility in soil

no data available

## 12.5 Other adverse effects

no data available

# **SECTION 13: Disposal considerations**

# 13.1 Disposal methods

#### Product

The material can be disposed of by removal to a licensed chemical destruction plant or by controlled incineration with flue gas scrubbing. Do not contaminate water, foodstuffs, feed or seed by storage or disposal. Do not discharge to sewer systems.

## Contaminated packaging

Containers can be triply rinsed (or equivalent) and offered for recycling or reconditioning. Alternatively, the packaging can be punctured to make it unusable for other purposes and then be disposed of in a sanitary landfill. Controlled incineration with flue gas scrubbing is possible for combustible packaging materials.

# **SECTION 14: Transport information**

#### 14.1 UN Number

ADR/RID: UN3115 (For reference only, please IMDG: UN3115 (For reference only, please theck.)

IATA: UN3115 (For reference only, please check.)

# 14.2 UN Proper Shipping Name

ADR/RID: ORGANIC PEROXIDE TYPE D, LIQUID, TEMPERATURE CONTROLLED (For reference only, please check.)

IMDG: ORGANIC PEROXIDE TYPE D, LIQUID, TEMPERATURE CONTROLLED (For reference only, please check.)

IATA: ORGANIC PEROXIDE TYPE D, LIQUID, TEMPERATURE CONTROLLED (For reference only, please check.)

## 14.3 Transport hazard class(es)

ADR/RID: 5.2 (For reference only, please check.)

IMDG: 5.2 (For reference only, please check.)

IATA: 5.2 (For reference only, please check.)

# 14.4 Packing group, if applicable

ADR/RID: (For reference only, please check.) IMDG: (For reference only, please check.) IATA: (For reference only, please check.)

#### 14.5 Environmental hazards

ADR/RID: No IMDG: No IATA: No

# 14.6 Special precautions for user

no data available

## 14.7 Transport in bulk according to IMO instruments

no data available

# **SECTION 15: Regulatory information**

# 15.1 Safety, health and environmental regulations specific for the product in question

Chemical name	Common names and synonyms	CAS number	EC number
Diacetyl peroxide	Diacetyl peroxide	110-22-5	203-748-8
European Inventory of Existing Commercial Chemical Substances (EINECS)			
EC Inventory			
United States Toxic Substances Control Act (TSCA) Inventory			
China Catalog of Hazardous chemicals 2015			
New Zealand Inventory of Chemicals (NZIoC)			
Philippines Inventory of Chemicals and Chemical Substances (PICCS)			
Vietnam National Chemical Inventory			
Chinese Chemical Inventory of Existing Chemical Substances (China IECSC)			
Korea Existing Chemicals List (KECL)			

## **SECTION 16: Other information**

#### Information on revision

**Creation Date** July 15, 2019 **Revision Date** July 15, 2019

#### Abbreviations and acronyms

- CAS: Chemical Abstracts Service
- ADR: European Agreement concerning the International Carriage of Dangerous Goods by Road
- RID: Regulation concerning the International Carriage of Dangerous Goods by Rail
- IMDG: International Maritime Dangerous Goods
- IATA: International Air Transportation Association
- TWA: Time Weighted Average
- STEL: Short term exposure limit
- LC50: Lethal Concentration 50%
- LD50: Lethal Dose 50%
- EC50: Effective Concentration 50%

#### References

- IPCS The International Chemical Safety Cards (ICSC), website: http://www.ilo.org/dyn/icsc/showcard.home
- HSDB Hazardous Substances Data Bank, website: https://toxnet.nlm.nih.gov/newtoxnet/hsdb.htm
- IARC International Agency for Research on Cancer, website: http://www.iarc.fr/
- eChemPortal The Global Portal to Information on Chemical Substances by OECD, website: http://www.echemportal.org/echemportal/index?pageID=0&request\_locale=en
- CAMEO Chemicals, website: http://cameochemicals.noaa.gov/search/simple
- ChemIDplus, website: http://chem.sis.nlm.nih.gov/chemidplus/chemidlite.jsp
- ERG Emergency Response Guidebook by U.S. Department of Transportation, website: http://www.phmsa.dot.gov/hazmat/library/erg
- Germany GESTIS-database on hazard substance, website: http://www.dguv.de/ifa/gestis/gestis-stoffdatenbank/index-2.jsp
- ECHA European Chemicals Agency, website: https://echa.europa.eu/