### **SECTION 1: Identification**

#### 1.1 GHS Product identifier

**Product name** 1,2-divinylbenzene

#### 1.2 Other means of identification

Product number

Other names Benzene, 1,2-diethenyl-; divinyl zinc; vinylstyrene

#### 1.3 Recommended use of the chemical and restrictions on use

Identified uses Intermediates, Ion exchange agents, Processing aids, specific to petroleum production

Uses advised against no data available

### **SECTION 2: Hazard identification**

#### 2.1 Classification of the substance or mixture

no data available

## 2.2 GHS label elements, including precautionary statements

Pictogram(s)no data availableSignal wordno data availableHazard statement(s)no data available

**Precautionary statement(s)** 

Preventionno data availableResponseno data availableStorageno data availableDisposalno data available

### 2.3 Other hazards which do not result in classification

no data available

## **SECTION 3: Composition/information on ingredients**

#### 3.1 Substances

Chemical name	Common names and synonyms	CAS number	EC number	Concentration
1,2-divinylbenzene	1,2-divinylbenzene	91-14-5	202-043-2	100%

## **SECTION 4: First-aid measures**

## 4.1 Description of necessary first-aid measures

#### If inhaled

Fresh air, rest.

#### Following skin contact

Remove contaminated clothes. Rinse skin with plenty of water or shower.

#### Following eye contact

First rinse with plenty of water for several minutes (remove contact lenses if easily possible), then refer for medical attention.

#### **Following ingestion**

Rinse mouth. Do NOT induce vomiting. Rest.

### 4.2 Most important symptoms/effects, acute and delayed

Exposure Routes: inhalation, ingestion, skin and/or eye contact Symptoms: irritation eyes, skin, respiratory system; skin burns; in animals: central nervous system depressant/depression Target Organs: Eyes, skin, respiratory system, central nervous system (NIOSH, 2016)

#### 4.3 Indication of immediate medical attention and special treatment needed, if necessary

Basic treatment: Establish a patent airway. Suction if necessary. Watch for signs of respiratory insufficiency and assist ventilations if necessary. Administer oxygen by nonrebreather mask at 10 to 15 L/min. Monitor for pulmonary edema and treat if necessary . Monitor for shock and treat if necessary . For eye contamination, flush eyes immediately with water. Irrigate each eye continuously with normal saline during transport . Do not use emetics. For ingestion, rinse mouth and administer 5 ml/kg up to 200 ml of water for dilution if the patient can swallow, has a strong gag reflex, and does not drool. Administer activated charcoal . Aromatic hydrocarbons and related compounds

## **SECTION 5: Fire-fighting measures**

## 5.1 Suitable extinguishing media

Use water spray, dry chemical, foam, or carbon dioxide. Use water spray to keep fire-exposed containers cool. Fight fire from protected lequipmentor maximum possible distance. Use remote equiment wherever possible.

### 5.2 Specific hazards arising from the chemical

This chemical is combustible. (NTP, 1992)

### 5.3 Special protective actions for fire-fighters

Wear self-contained breathing apparatus for firefighting if necessary.

#### **SECTION 6: Accidental release measures**

#### 6.1 Personal precautions, protective equipment and emergency procedures

Avoid dust formation. Avoid breathing mist, gas or vapours. Avoid contacting with skin and eye. Use personal protective equipment. Wear chemical impermeable gloves. Ensure adequate ventilation. Remove all sources of ignition. Evacuate personnel to safe areas. Keep people away from and upwind of spill/leak.

#### 6.2 Environmental precautions

Personal protection: filter respirator for organic gases and vapours adapted to the airborne concentration of the substance. Collect leaking and spilled liquid in sealable containers as far as possible. Absorb remaining liquid in sand or inert absorbent. Then store and dispose of according to local regulations.

#### 6.3 Methods and materials for containment and cleaning up

Stop or control the leak, if this can be done without undue risk. Adsorb in noncombustible material (e.g., sand) for proper disposal. Use water spray to cool and disperse vapors and protect personnel. Control runoff and isolate discharged material for proper disposal.

## **SECTION 7: Handling and storage**

#### 7.1 Precautions for safe handling

Handling in a well ventilated place. Wear suitable protective clothing. Avoid contact with skin and eyes. Avoid formation of dust and aerosols. Use non-sparking tools. Prevent fire caused by electrostatic discharge steam.

### 7.2 Conditions for safe storage, including any incompatibilities

Separated from oxidants. Cool. Store only if stabilized. Store in a cool, dry, well-ventilated location. Outside or detached storage is preferred.

## **SECTION 8: Exposure controls/personal protection**

### 8.1 Control parameters

#### Occupational Exposure limit values

Component	1,2-divinylbenzene				
CAS No.	91-14-5				
	Limit value - Eight hours		Limit value - Short term		
	ppm	mg/m <sup>3</sup>	ppm	mg/m <sup>3</sup>	
Denmark	10	50	20	100	
	Remarks				

#### **Biological limit values**

no data available

## 8.2 Appropriate engineering controls

Ensure adequate ventilation. Handle in accordance with good industrial hygiene and safety practice. Set up emergency exits and the risk-elimination area.

## 8.3 Individual protection measures, such as personal protective equipment (PPE)

### Eye/face protection

Wear tightly fitting safety goggles with side-shields conforming to EN 166(EU) or NIOSH (US).

### Skin protection

Wear fire/flame resistant and impervious clothing. Handle with gloves. Gloves must be inspected prior to use. Wash and dry hands. The selected protective gloves have to satisfy the specifications of EU Directive 89/686/EEC and the standard EN 374 derived from it.

#### Respiratory protection

If the exposure limits are exceeded, irritation or other symptoms are experienced, use a full-face respirator.

#### Thermal hazards

no data available

Physical state Divinyl benzene is a water-white to straw colored liquid. Slightly less dense than water and

insoluble in water. Vapors may be toxic. Used in making rubber.

**Colour** Pale, straw-colored liquid.

Odourno data availableMelting point/freezing point-125° F (NTP, 1992)Boiling point or initial boiling point207.3°C at 760 mmHg

and boiling range

**Flammability** Class IIIA Combustible Liquid: Fl.P. at or above 140°F and below 200°F. **Lower and upper explosion** Lower flammable limit: 0.7% by volume; Upper flammable limit: 6.2% by volume

limit/flammability limit

Flash point 71.9°C

Auto-ignition temperature 500 deg C

Decomposition temperature no data available pH no data available in data available no data available no data available

**Solubility** less than 1 mg/mL at  $64\text{Å}^{\circ}$  F (NTP, 1992)

**Partition coefficient n-octanol/water** log Kow = 3.8 /Estimated/ **Vapour pressure** 0.7 mm Hg (NIOSH, 2016)

Density and/or relative density0.921g/cm3Relative vapour density(air = 1): 4.48Particle characteristicsno data available

# **SECTION 10: Stability and reactivity**

### 10.1 Reactivity

The substance may polymerize due to heating. This generates fire or explosion hazard. Reacts violently with oxidants.

## 10.2 Chemical stability

no data available

### 10.3 Possibility of hazardous reactions

Combustible. Heating will cause rise in pressure with risk of bursting.DIVINYL BENZENE may react vigorously with strong oxidizing agents. Can react exothermically with reducing agents (such as alkali metals and hydrides) to release gaseous hydrogen. May react exothermically with both acids and bases. May in the presence of various catalysts (such as acids) or initiators undergo exothermic polymerization. Inhibited by presence of an additive. When uninhibited violent polymerization may occur (NTP, 1992). Substitution at the benzene nucleus occurs by halogenation (acid catalyst), nitration, sulfonation, and the Friedel-Crafts reaction.

### 10.4 Conditions to avoid

no data available

### 10.5 Incompatible materials

Reacts with acids, oxidizing materials, peroxides, metal salts such as iron chloride or aluminum chloride, or polymer initiators.

## 10.6 Hazardous decomposition products

When heated to decomposition it emits acrid smoke and irritating fumes

## **SECTION 11: Toxicological information**

#### Acute toxicity

- Oral: LD50 Rat (male) oral 4640 mg/kg
- Inhalation: no data available
- Dermal: no data available

### Skin corrosion/irritation

no data available

### Serious eye damage/irritation

no data available

### Respiratory or skin sensitization

no data available

### Germ cell mutagenicity

no data available

### Carcinogenicity

no data available

#### Reproductive toxicity

no data available

#### STOT-single exposure

no data available

#### STOT-repeated exposure

no data available

#### **Aspiration hazard**

no data available

## **SECTION 12: Ecological information**

### 12.1 Toxicity

- Toxicity to fish: no data available
- Toxicity to daphnia and other aquatic invertebrates: no data available
- Toxicity to algae: no data available
- Toxicity to microorganisms: no data available

### 12.2 Persistence and degradability

In a 4-week biodegradation screening test (MITI test) using vinylstyrene (100 mg/l) and an activated sludge inoculum (30 mg/l), 0% of BOD was removed(1). Data regarding the biodegradation of vinylstyrene in either soil or aquatic systems were not available(SRC).

### 12.3 Bioaccumulative potential

A BCF range of 206-444 was measured for vinylstyrene(1). According to a classification scheme(2), these BCF values suggest the potential for bioconcentration in aquatic organisms is high(SRC).

#### 12.4 Mobility in soil

Using a structure estimation method based on molecular connectivity indices(1), the Koc for vinylstyrene can be estimated to be 1,700(SRC). According to a classification scheme(2), this estimated Koc value suggests that vinylstyrene is expected to have low mobility in soil.

#### 12.5 Other adverse effects

no data available

## **SECTION 13: Disposal considerations**

### 13.1 Disposal methods

#### Product

The material can be disposed of by removal to a licensed chemical destruction plant or by controlled incineration with flue gas scrubbing. Do not contaminate water, foodstuffs, feed or seed by storage or disposal. Do not discharge to sewer systems.

### Contaminated packaging

Containers can be triply rinsed (or equivalent) and offered for recycling or reconditioning. Alternatively, the packaging can be punctured to make it unusable for other purposes and then be disposed of in a sanitary landfill. Controlled incineration with flue gas scrubbing is possible for combustible packaging materials.

## **SECTION 14: Transport information**

### 14.1 UN Number

ADR/RID: no data available IMDG: no data available IATA: no data available

### 14.2 UN Proper Shipping Name

ADR/RID: no data available IMDG: no data available IATA: no data available

### 14.3 Transport hazard class(es)

ADR/RID: no data available IMDG: no data available IATA: no data available

### 14.4 Packing group, if applicable

ADR/RID: no data available IMDG: no data available IATA: no data available

#### 14.5 Environmental hazards

ADR/RID: No IMDG: No IATA: No

### 14.6 Special precautions for user

no data available

# **SECTION 15: Regulatory information**

## 15.1 Safety, health and environmental regulations specific for the product in question

Chemical name	Common names and synonyms	CAS number	EC number
1,2-divinylbenzene	1,2-divinylbenzene	91-14-5	202-043-2
European Inventory of Existing Commercial Chemical Substances (EINECS)			
EC Inventory			Listed.
United States Toxic Substances Control Act (TSCA) Inventory			
China Catalog of Hazardous chemicals 2015			
New Zealand Inventory of Chemicals (NZIoC)			
Philippines Inventory of Chemicals and Chemical Substances (PICCS)			
Vietnam National Chemical Inventory			
Chinese Chemical Inventory of Existing Chemical Substances (China IECSC)			
Korea Existing Chemicals List (KECL)			

## **SECTION 16: Other information**

#### Information on revision

**Creation Date** July 15, 2019 **Revision Date** July 15, 2019

#### Abbreviations and acronyms

- CAS: Chemical Abstracts Service
- ADR: European Agreement concerning the International Carriage of Dangerous Goods by Road
- RID: Regulation concerning the International Carriage of Dangerous Goods by Rail
- IMDG: International Maritime Dangerous Goods
- IATA: International Air Transportation Association
- TWA: Time Weighted Average
- STEL: Short term exposure limit
- LC50: Lethal Concentration 50%
- LD50: Lethal Dose 50%
- EC50: Effective Concentration 50%

### References

- IPCS The International Chemical Safety Cards (ICSC), website: http://www.ilo.org/dyn/icsc/showcard.home
- HSDB Hazardous Substances Data Bank, website: https://toxnet.nlm.nih.gov/newtoxnet/hsdb.htm
- IARC International Agency for Research on Cancer, website: http://www.iarc.fr/
- eChemPortal The Global Portal to Information on Chemical Substances by OECD, website: http://www.echemportal.org/echemportal/index?pageID=0&request\_locale=en
- $\bullet \quad \text{CAMEO Chemicals, website: http://cameochemicals.noaa.gov/search/simple} \\$
- ChemIDplus, website: http://chem.sis.nlm.nih.gov/chemidplus/chemidlite.jsp
- ERG Emergency Response Guidebook by U.S. Department of Transportation, website: http://www.phmsa.dot.gov/hazmat/library/erg
- Germany GESTIS-database on hazard substance, website: http://www.dguv.de/ifa/gestis/gestis-stoffdatenbank/index-2.jsp
- ECHA European Chemicals Agency, website: https://echa.europa.eu/